

# Effect of surface fluorination on GaN deep dry etching defects

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PESM2014  
May 12<sup>th</sup> 2014

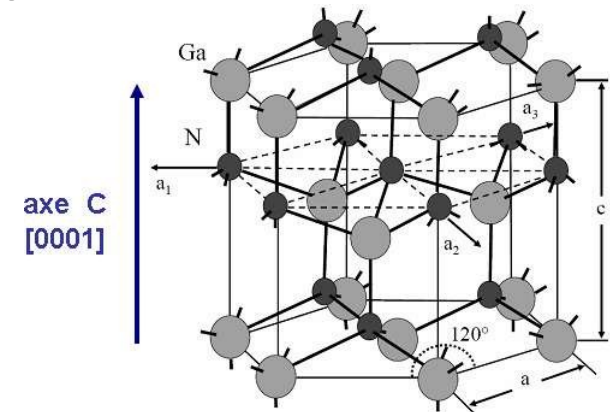
# Outline

- I. Introduction
- II. Experimental set-up
- III. Study of fluorine addition in etching chemistries
- IV. Conclusion

# Introduction

## Gallium nitride

- ▶ Gallium nitride (GaN) = III-V semiconductor
- ▶ GaN = direct and wide bandgap (3.4 eV)
- ▶ Hexagonal structure (Wurtzite)
- ▶ Interesting properties for power devices :
  - High breakdown voltage
  - High electron mobility
  - High saturation velocity
- ▶ Heteroepitaxial growth along c-plane on Silicon or Sapphire substrates  $\Rightarrow$  Defects (dislocations, vacancies, nanopipes)



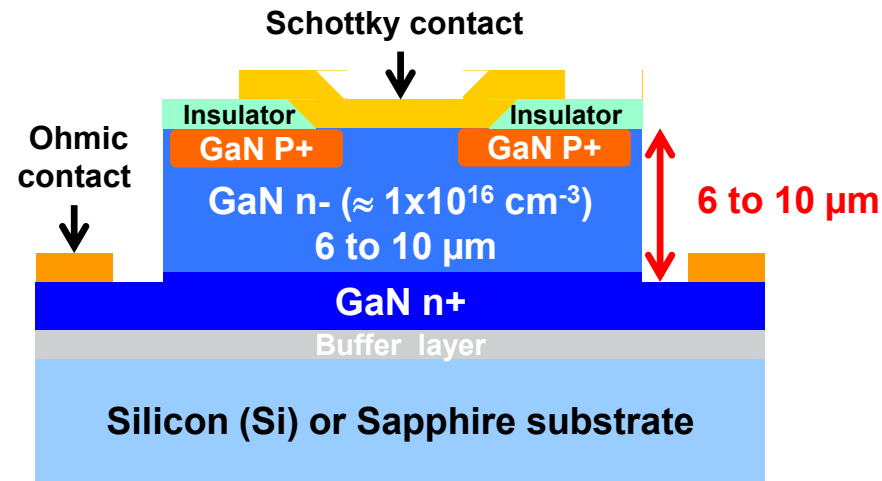
# Introduction

## Schottky diode structure

- ▶ Aim : Alternating current to direct current conversion

⇒ GaN based Schottky diode (600 V)

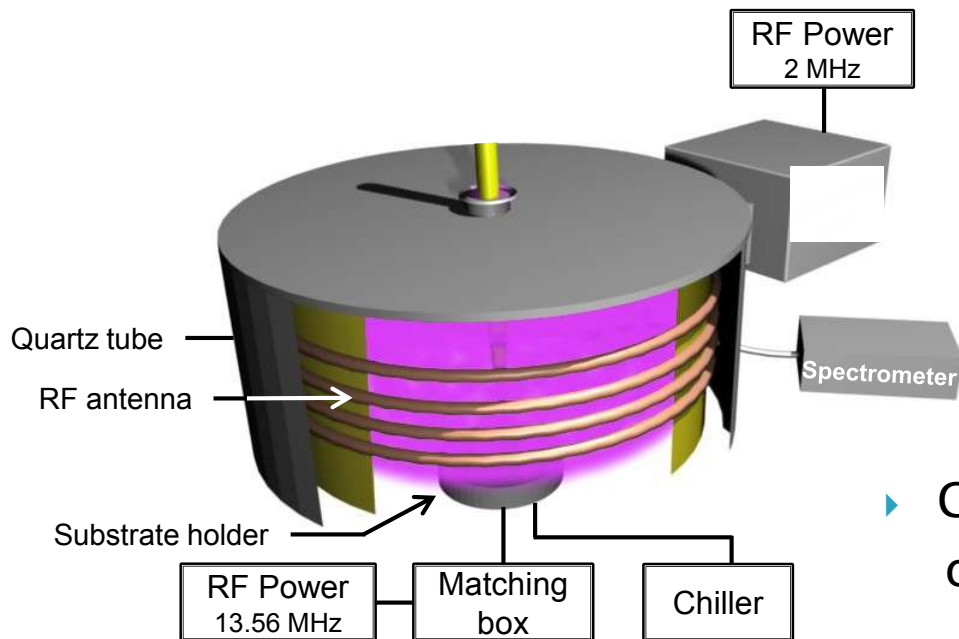
- ▶ Pseudo-vertical structure :



- ▶ Deep etching of GaN is needed to perform mesa structure
  - ▶ Wet etching does not work on GaN c-plane ⇒ Plasma etching
    - Etch rate and selectivity
    - Surface defects and roughness
- } optimization

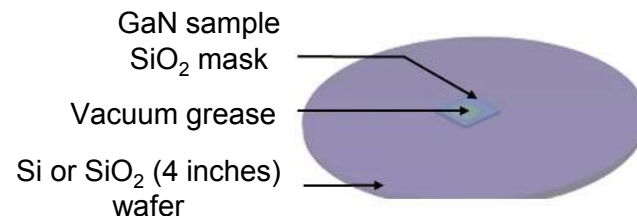
# Experimental set-up

- ▶ GaN etching is performed with an ICP reactors Corial 200-IL



- ICP source
- Without diffusion chamber
- High bias voltage

- ▶ Etching on samples :



⇒ Importance of the carrier wafer composition

- ▶  $\text{GaCl}_3$  is the most volatile etch product of GaN

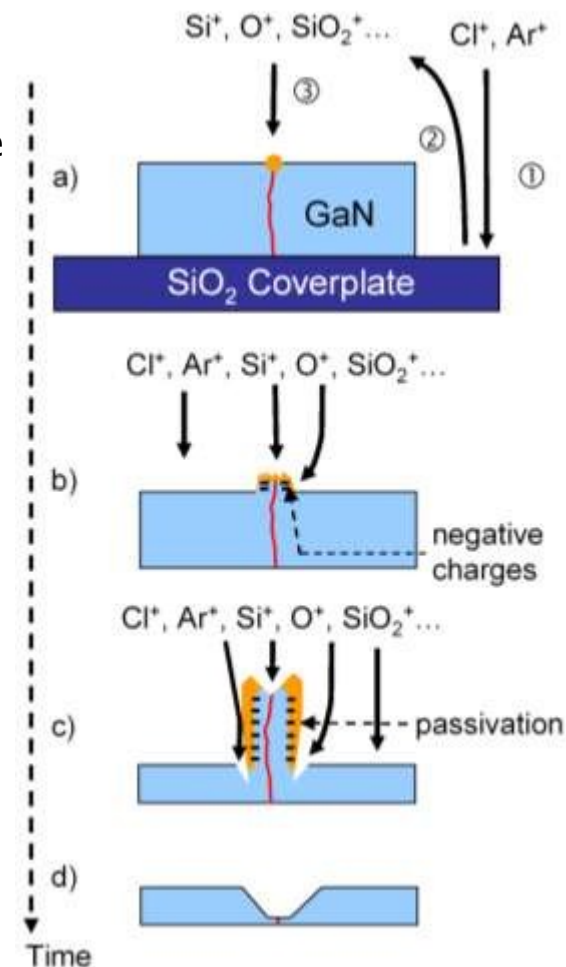
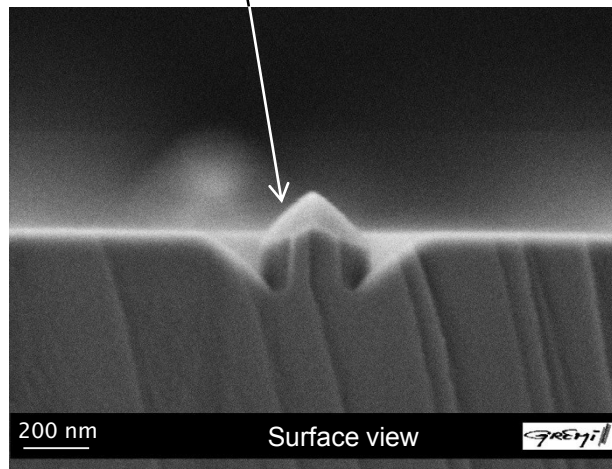
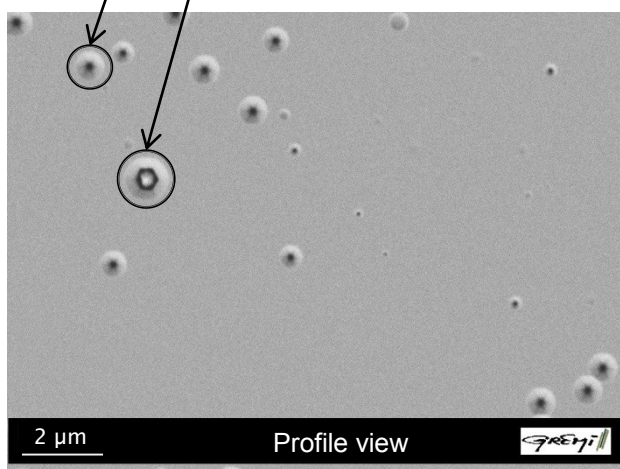
⇒ Chlorine-based chemistries are used for etching

- ▶ Addition of Argon ( $\text{Ar}^+$  ions) to enhance the desorption of volatile products and GaN etch rate

⇒ Study of GaN etching in  $\text{Cl}_2/\text{Ar}$  based plasma

# GaN etching and defect formation

- ▶ 2 types of defects on GaN etch surface :
  - Cavity defects
    - Due to the ion bombardment
    - Faceting with  $\text{Cl}_2$  based etching  $\Rightarrow$  hexagonal structure
  - Columnar defects
    - Passivation of nanopipes present in the bulk material
    - Important role of oxygen species

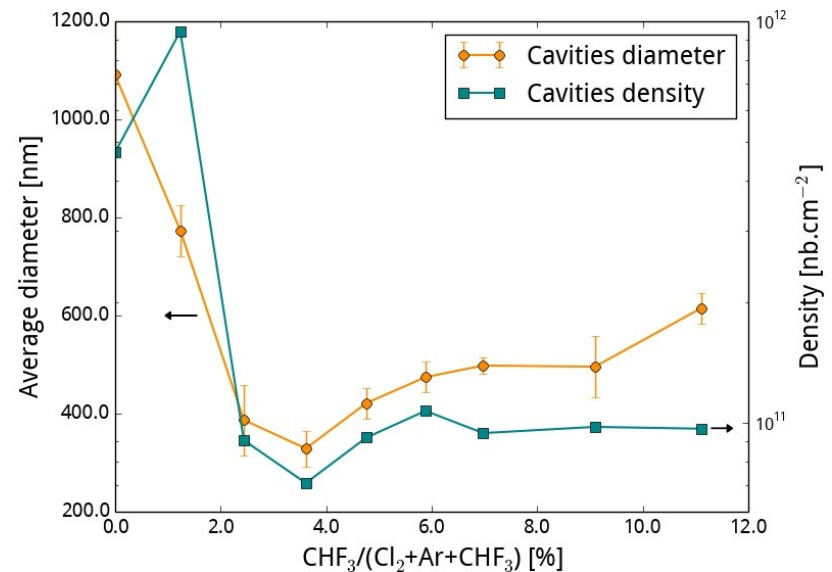
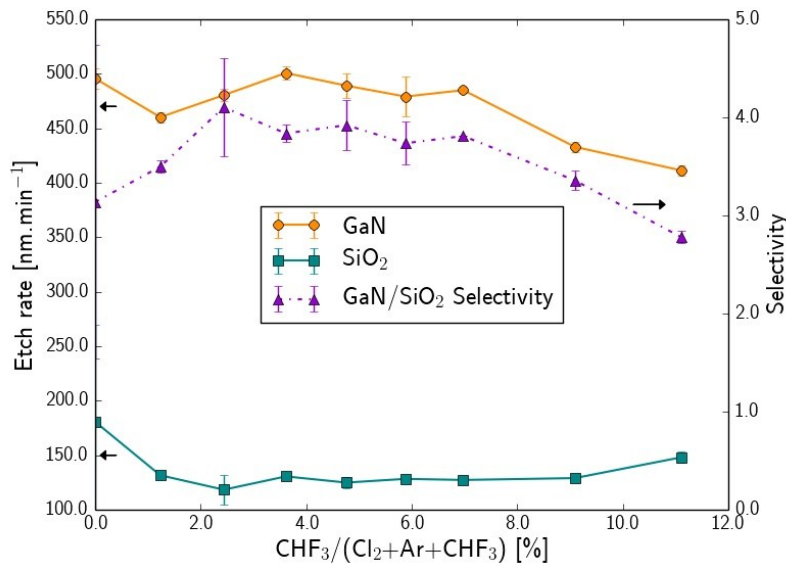


# Study of fluorine addition in GaN etching chemistries

## Addition of CHF<sub>3</sub> in Cl<sub>2</sub>/Ar



- ▶ As shown previously, formation of columnar defects is related to oxygen species
- ⇒ Addition of CHF<sub>3</sub> in Cl<sub>2</sub>/Ar chemistry in order to scavenge oxygen impurities
- ▶ The etch time was 5 minutes for all the following experiments



From 2 to 5 % of CHF<sub>3</sub> in Cl<sub>2</sub>/Ar : – GaN Etch rate ≈ constant  
 – Diameter and density of defects ↘  
 – No (or few) columnar defects

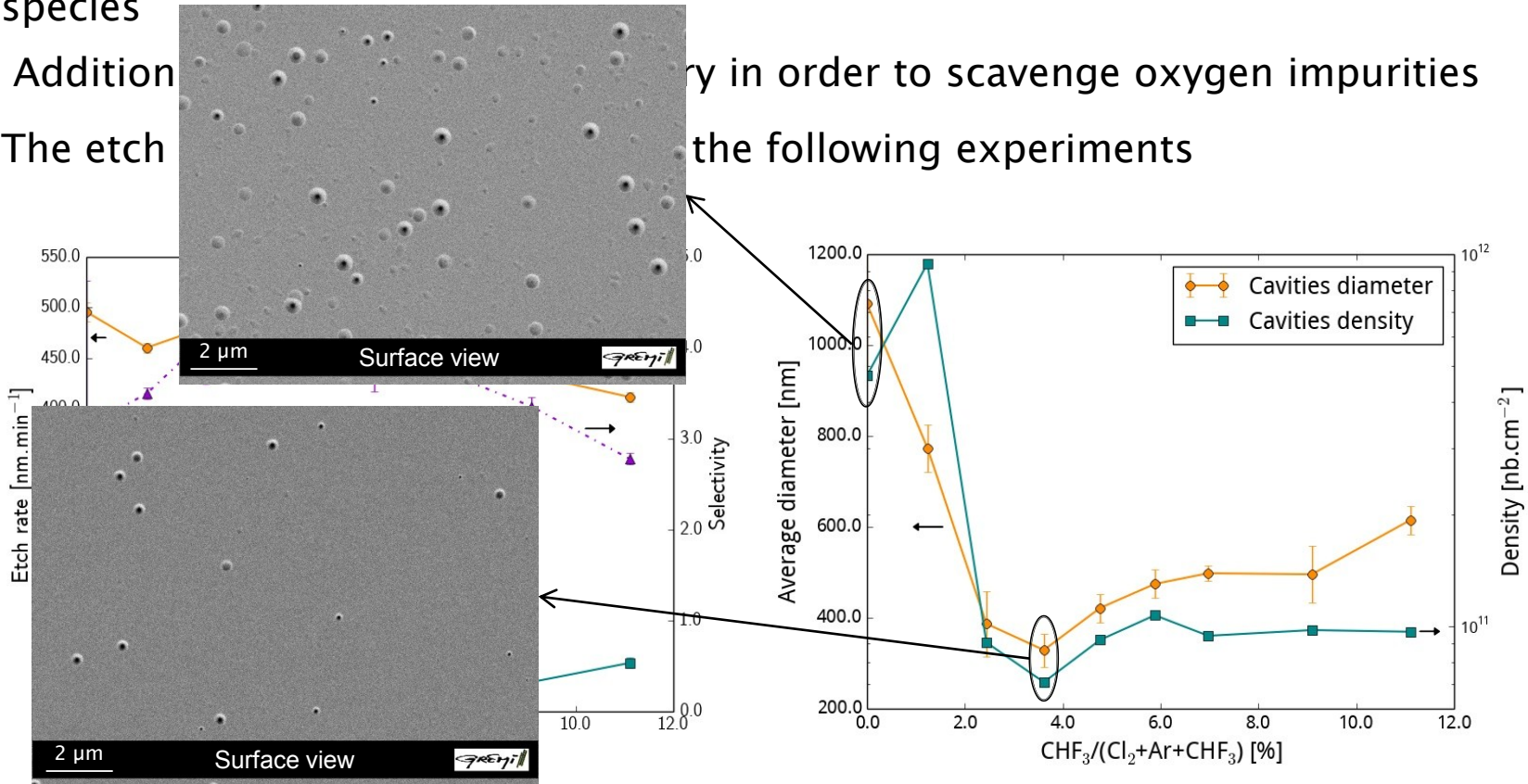
⇒ Impact of fluorine species ?

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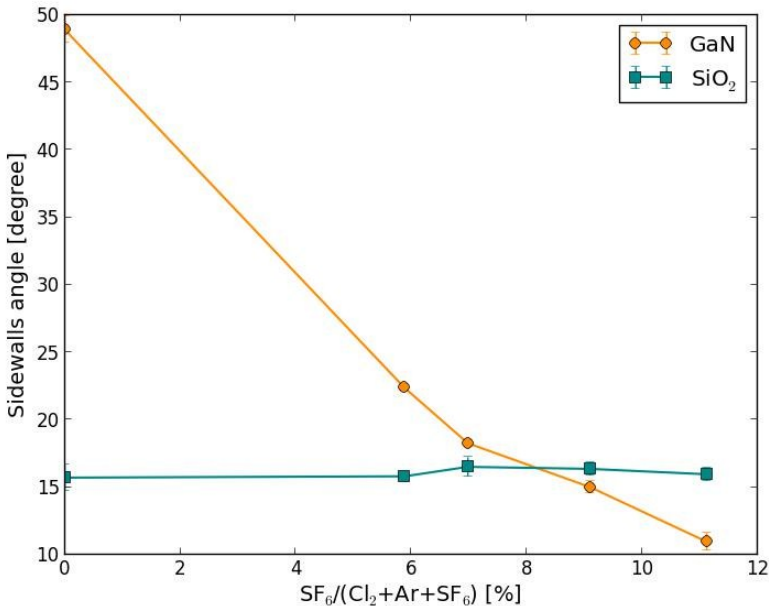
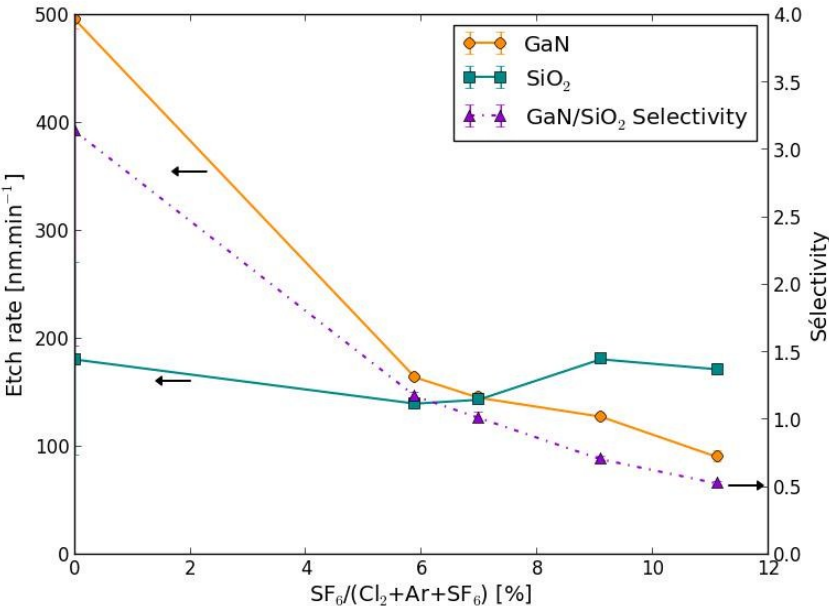


# Study of fluorine addition in GaN etching chemistries

## Addition of SF<sub>6</sub> in Cl<sub>2</sub>/Ar



- ▶ SF<sub>6</sub> = - More fluorine radicals than in CHF<sub>3</sub>
- No carbon able to react with GaN and no CF<sub>x</sub> passivation layer deposition



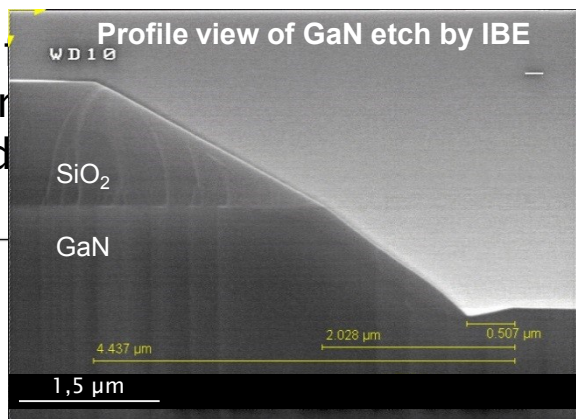
- With SF<sub>6</sub> ↑ in Cl<sub>2</sub>/Ar:
    - GaN etch rate ↓
    - Sidewalls angle of GaN ↓
  - No defects observed !
- } Looks-like pure physical etching (IBE)

# Study of fluorine addition in GaN etching chemistries

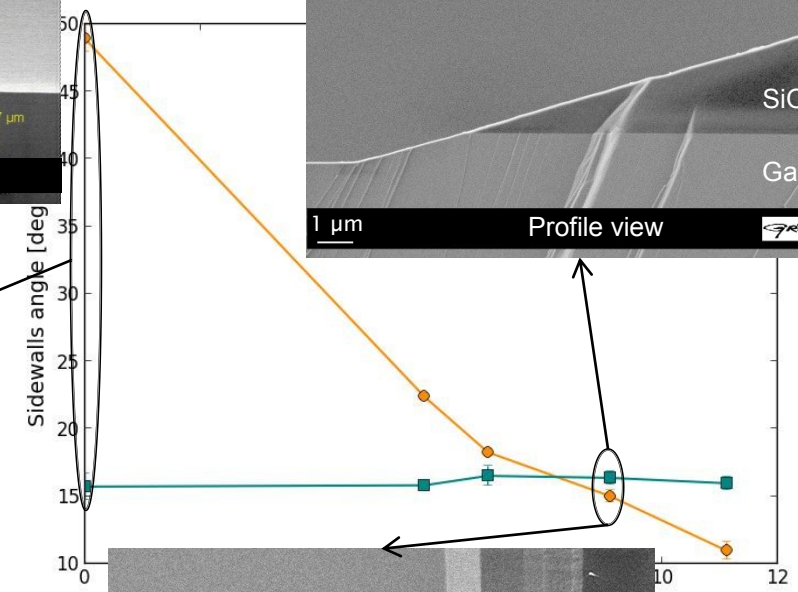
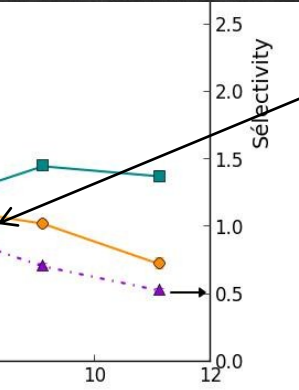
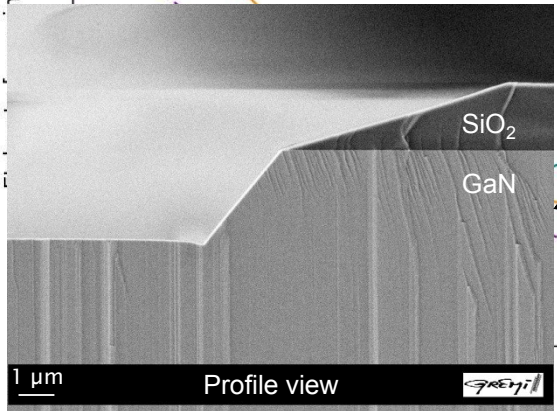
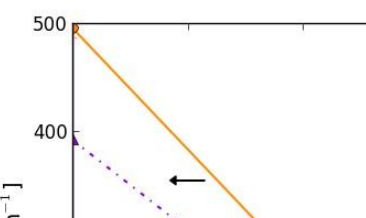
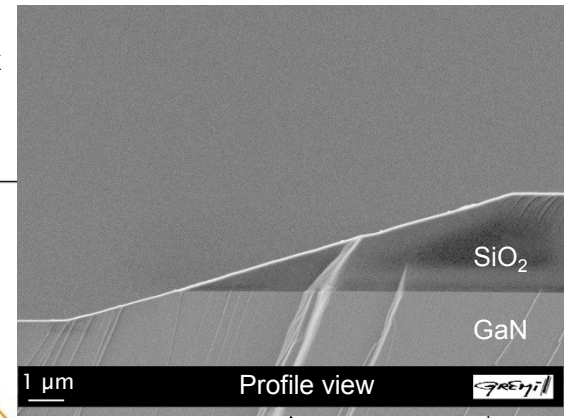
## Addition of SF<sub>6</sub> in Cl<sub>2</sub>/Ar



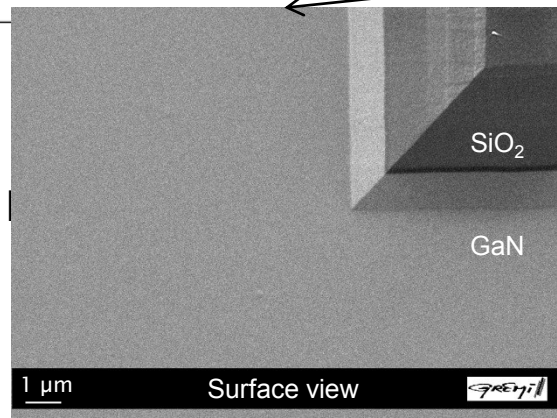
- ▶ SF<sub>6</sub> = - More
- No car
- layer d



and no CF<sub>x</sub>



- With SF<sub>6</sub> ↑ in Cl<sub>2</sub>/Ar:
  - GaN etch rate ↓
  - Sidewalls angle of GaN ↓
- No defects observed !



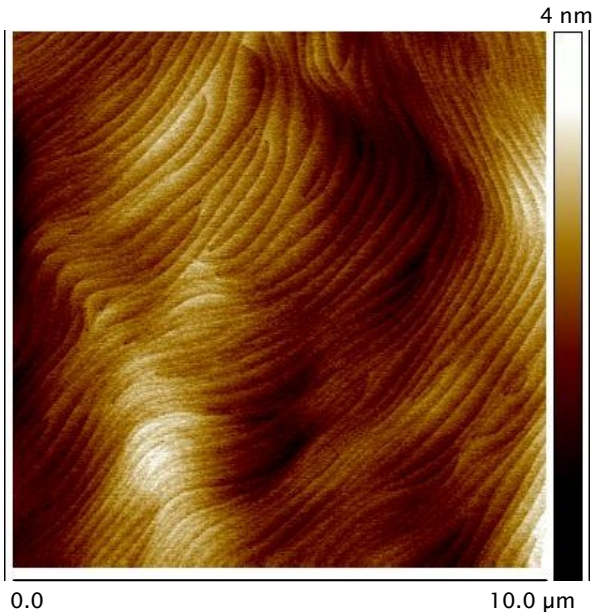
etching (IBE)

# Study of fluorine addition in GaN etching chemistries

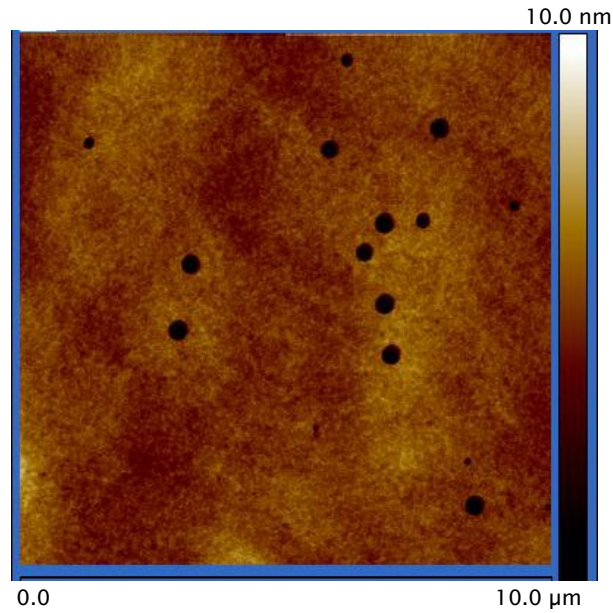
## Comparison between $\text{CHF}_3$ and $\text{SF}_6$



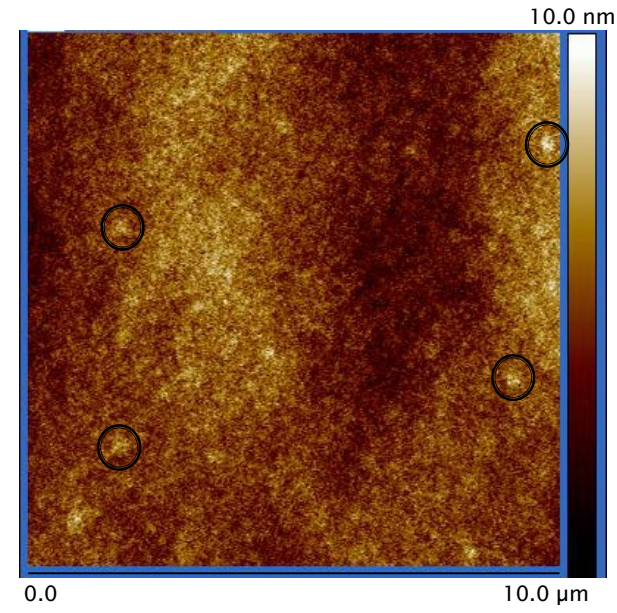
- ▶ AFM measurements before and after etching :



**Before GaN etching**  
**RMS = 0.181 nm**



**After etching with 2 sccm  $\text{CHF}_3$**   
**RMS = 0.378 nm**



**After etching with 6 sccm  $\text{SF}_6$**   
**RMS = 0.822 nm**

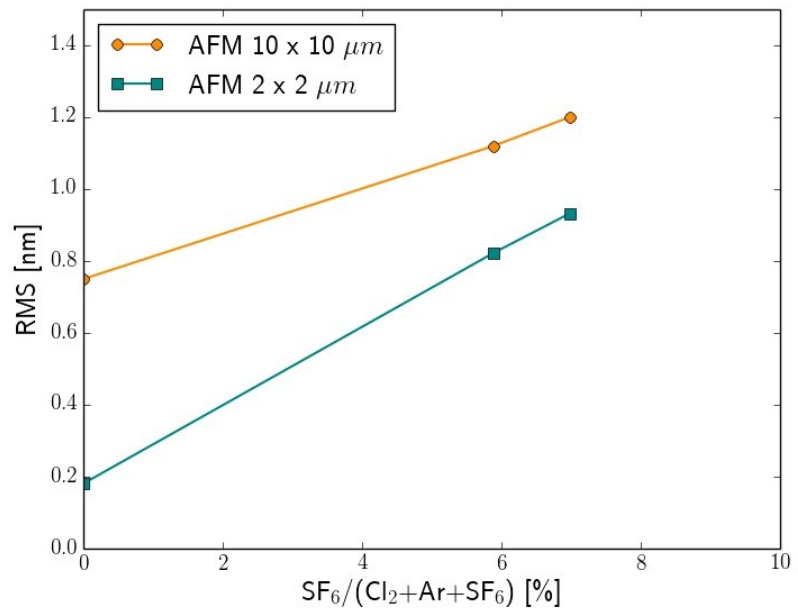
- Surface roughness is more significant when etched with  $\text{SF}_6$
- Surface morphology is different with addition of  $\text{SF}_6$  (white dots)

# Study of fluorine addition in GaN etching chemistries

## Comparison between CHF<sub>3</sub> and SF<sub>6</sub>



AFM measurements



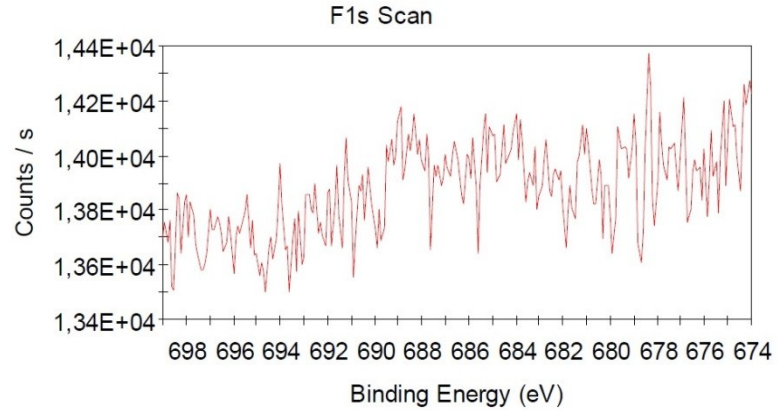
- ▶ When SF<sub>6</sub> flow ↑ RMS roughness of GaN ↑
- ▶ Fluorine detected only on GaN etched surface with Cl<sub>2</sub>/Ar/SF<sub>6</sub> chemistry

⇒ Formation of Ga<sub>x</sub>N<sub>y</sub>F<sub>z</sub> layer

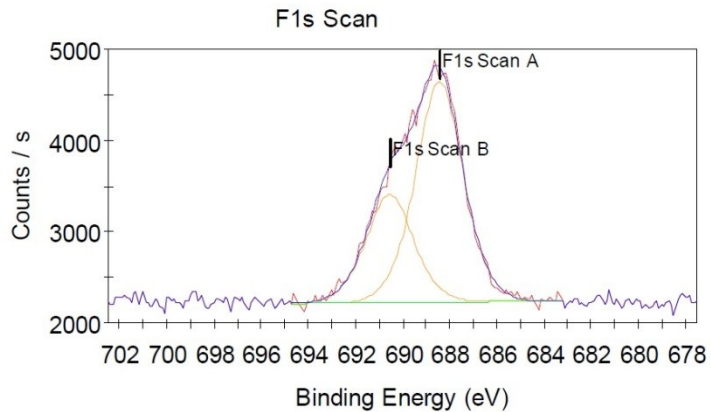


No defects on the etched surface

XPS measurements



XPS spectra on GaN etch with Cl<sub>2</sub>/Ar/CHF<sub>3</sub> chemistry



XPS spectra on GaN etch with Cl<sub>2</sub>/Ar/SF<sub>6</sub> chemistry

# Conclusion

- ▶ GaN etching in  $\text{Cl}_2/\text{Ar}$  chemistry :
    - High etch rate is possible
    - 2 types of defects are revealed : cavities and columns
  
  - ▶  $\text{CHF}_3$  addition in  $\text{Cl}_2/\text{Ar}$  :
    - Important reduction of the density and diameter of the cavities
    - No columnar defects
    - Constant GaN etch rate
  
  - ▶  $\text{SF}_6$  addition in  $\text{Cl}_2/\text{Ar}$  :
    - Important decrease of GaN etch rate
    - Important surface roughness
    - Fluorine species in GaN surface (XPS)
- Formation of  $\text{Ga}_x\text{N}_y\text{F}_z$   
 ⇒ No defects on GaN etch surface

Thank you for your attention